

SIMPLE SYNTHESIS OF 2-ACETYL-5,8-DIMETHOXY-3,4-DIHYDRONAPHTHALENE, A KEY  
 INTERMEDIATE FOR THE SYNTHESIS OF OPTICALLY ACTIVE ANTHRACYCLONES

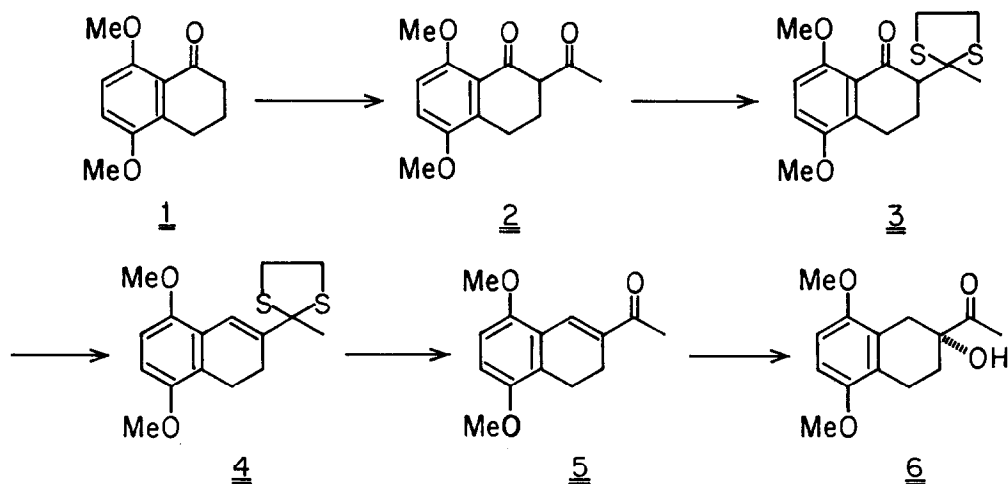
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*2-Acetyl-5,8-dimethoxy-3,4-dihydronaphthalene is synthesised easily starting from 2-acetyl-5,8-dimethoxytetralone.*

Recent communication<sup>1</sup> on the synthesis of 2-acetyl-5,8-dimethoxy-3,4-dihydronaphthalene (5) had prompted us to report our findings on its synthesis which was completed some time back. Earlier, we had reported a simple synthesis of (+) 2-acetyl-2-hydroxy-5,8-dimethoxytetralin (6) which was then elaborated to (+)-4-demethoxydaunomycinone.<sup>2</sup> Compound 6 was originally synthesised in nine steps starting from 2,5-dimethoxybenzaldehyde<sup>3</sup> and the optically active (R)(-)-6 has been achieved by optical resolution.<sup>4</sup> Further synthetic routes from (R)(-)-6 to optical active anthracyclones have been established.<sup>5</sup> An elegant and most rational approach for the synthesis of (R)(-)-6 making use of 2-acetyl-5,8-dimethoxy-3,4-dihydronaphthalene (5) as a key intermediate has been reported recently by Terashima *et al.*<sup>6</sup> However, their synthesis of 5 is more tedious and not convenient for the preparation of 6 in gram quantities. We have achieved the synthesis of 5 in much shorter route and in better yields.

Our synthetic approach is shown in scheme I. 2-Acetyl-5,8-dimethoxytetralone (2) is made from 1 either by condensing with sodium and ethyl acetate or  $\text{BF}_3$  etherate- $\text{Ac}_2\text{O}$ <sup>7</sup> in 85% yield. The acetyl ketone group was then selectively protected ( $\text{HS-CH}_2\text{CH}_2\text{-SH}$ ,  $\text{HCl}$  gas,  $\text{CHCl}_3$ , r.t., 10 hr) to give ketone 3<sup>8</sup>, [PMR,  $\text{CCl}_4$ ,  $\delta$  1.70 (s, 3H,  $\text{CH}_3$ ), 1.90-2.80 (m, 5H, CH and 2 x  $\text{CH}_2$ ), 2.90 (2, 4H,  $-\text{S}(\text{CH}_2)_2\text{S}-$ ), 3.46 (s, 6H, 2xOMe), 5.96, 6.16 (dd,  $J=9$  Hz, 2H, ArH)]. Compound 3 was subjected to reduction ( $\text{NaBH}_4$ , MeOH, r.t., 24 hr) followed by acid work up gave 4 in 76% yield, m.p. 95-97°, [PMR,  $\text{CCl}_4$ ,  $\delta$  2.0 (s, 3H,  $\text{CH}_3$ ), 2.53 (m, 4H,  $2\text{CH}_2$ ), 3.36 (s, 4H,  $-\text{S}(\text{CH}_2)_2\text{S}-$ ), 3.76 (s, 6H, 2xOMe), 6.43 (s, 2H, ArH), 6.83 (bs, 1H, vinylic C-H)]. Deketalization of 4 ( $\text{NCS}$ ,  $\text{AgNO}_3$ , 80% aq.  $\text{CH}_3\text{CN}$ , r.t., 20 min) gave the desired 2-acetyl-5,8-dimethoxy-3,4-dihydronaphthalene (5) in 73% yield, m.p. 102-103° (Lit.<sup>6,1</sup> m.p. 104-105°, m.p. 102-103°) [PMR,  $\text{CCl}_4$ ,  $\delta$  2.36, s, 3H,  $\text{CH}_3$ ), 2.60 (m, 4H,  $2\text{xCH}_2$ ), 3.76, 3.83 (2s, 6H, 2 x OMe), 7.53, 7.76 (dd,  $J=9$  Hz, 2H, ArH), 7.70 (bs, 1H, vinylic C-H)]<sup>9</sup>.

Scheme 1



This approach is being extended for the synthesis of optically active aklavinone, the aglycone of aclacinomycin A<sup>10,11</sup>.

#### References and Notes

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8. Protection of **2** with ethylene glycol, PTSA cleaves the acetyl group to give **1** ketal. Although partial thioketalization of **2** proceeds smoothly to give **3** but often the product is not free from **1** thioketal which can be removed from **3** by silica gel chromatography. The conditions for the exclusive formation of **3** are being optimised.
9. All the compounds gave satisfactory elemental analyses and spectroscopic data.
10. W.A. Remers "The chemistry of antitumor antibiotics" Vol.1, Wiley NY. 1979, p. 63-126.
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